

## STUDIES OF GENTAMYCIN ADSORPTION ON CHITOSAN

**Jan Meler, Bożena Grimling, Janusz Pluta**

*Department of Drug Form Technology,  
Medical University of Wrocław,  
ul. Borowska 211A, 50-556 Wrocław, Poland  
E-mail: jan.meler@am.wroc.pl*

### **Abstract**

*Chitosan as a dietary supplement has been used in the treatment of obesity due to its capability of binding bioactive compounds. A molecule of a cationic polymer, such as chitosan, may bind gentamycin. The aim of the study was to determine the binding capability of Gentamycin by chitosan's contained in weight-loss supplements. The Gentamycin absorption was investigated by means of a dynamic method in a biopharmaceutical model imitating in vitro conditions including the required level of nutrients. Mean absorption of gentamycin by chitosan ranged from 2.1% to 34.1% depending on the environmental pH. The highest absorption rate was observed at above pH 6.8. Concluding, the studies confirmed that the investigated gentamycin interact with chitosan, what leads to decrease in their amount and affects the bioavailability of the drugs.*

**Key words:** *Gentamycin, chitosan; absorption.*

## **1. Introduction**

The combinations of polymers and biopolymers with biologically active small-molecule compounds have become the subject of extensive investigations recently. A small molecular weight active substance when combined with a polymer often presents a modified action. On the other hand, the use of improper polymers may lead to drug-polymer incompatibility. Especially important are interactions in the form of adsorption and formation of complex bonds diminishing the effect of gentamycin. Chitosan, due to its capacity of binding lipids, cholesterol, fatty acids, triglycerides, bile acids, has been used in the treatment of obesity. It is an effective source of soluble fiber. It does not undergo digestion; it dissolves in the acid environment of the stomach, where it binds many molecules of water, forming a stable absorption gel. A molecule of a cationic polymer, such as chitosan, is capable of binding acid drugs [1, 2].

For this reason the study was undertaken to explain the mechanism of interactions of antibiotics drug: Gentamycin with dietary supplements containing chitosan. Qualitative study of adsorption will be conducted in future work.

## **2. Materials and method**

Natural chitosans ( Primex, Chito-Clear, Huasu) with deacetylation of 85% to 95%, degraded by 5 to 30 kGy radiation dose were used in the study. Also, dietary supplements containing chitosan were used (Vitana<sup>®</sup>, Hitec Nutrition<sup>®</sup>, Chromdiet<sup>®</sup>, Bio-Active Tech-Food Trading<sup>®</sup>). The adsorption of the gentamycin was investigated by means of a dynamic method in a biopharmaceutical model imitating the conditions *in vitro*. The amount of gentamycin absorbed by chitosan was calculated from the concentration of the investigated drug prior to and after sorption. The calculated amounts of bound drug were used to calculate mean percentage of absorbed dose. 0.03 g portions of chitosan end equivalent drug (0,060 mg) using were weighed and put to 5 ml glass centrifuge vials and next 2 ml of 0.05 M HCl were added to achieve pH 2 of the solution, what corresponds to natural fasting gastric pH [5 - 8].

Next the drugs were added to the vials: Gentamycin the doses of 0.01 g. Next 0.1 M Na<sub>2</sub>CO<sub>3</sub> was added to the vials to achieve pH 6.4 (corresponding to the duodenal pH) and stirred for 0.5 hour (300 r.p.m.). The solution was alkalized with sodium carbonate to achieve pH 7.0 - 7.6, corresponding to the pH of the intestinal juice. The samples were then incubated at 37 °C in a shaker (300 r.p.m.) for 2.5 hours, after which the samples were brought to room temperature and centrifuged (2100 ×g) for 20 minutes.

Then left at standstill in order to align the layers for a further 30 min. The mixture was filtered through filter paper, and the clear solution was diluted 1:1 with 0.05 M HCl using a spectrophotometric method were tested at a wavelength of 218 nm. The concentration of antibiotic in the test solution was calculated by interpolation of the results of absorbance with respect to the equation of a straight absorbance antibiotic concentration in the solution and the result obtained by multiplying the value of the sample used in the dilution.

Measurements were carried out on the basis of three independent sample, which lead to the arithmetic and assessed statistically. Calculation of standard deviation was made, and then determining the coefficient of relativity repeatability was calculated. Standard deviation ranged from 0.001 to 0.057 and the determined relativity coefficient ranged from 1.81% to 4.82%, what confirms high accuracy of measurements [9 - 12].

Research of chitosan viscosity used in the work was carried out by the procedure described in previous work [12].

### **3. Results and discussion**

#### **3.1. Gentamycin binding by degraded and non-degraded chitosans**

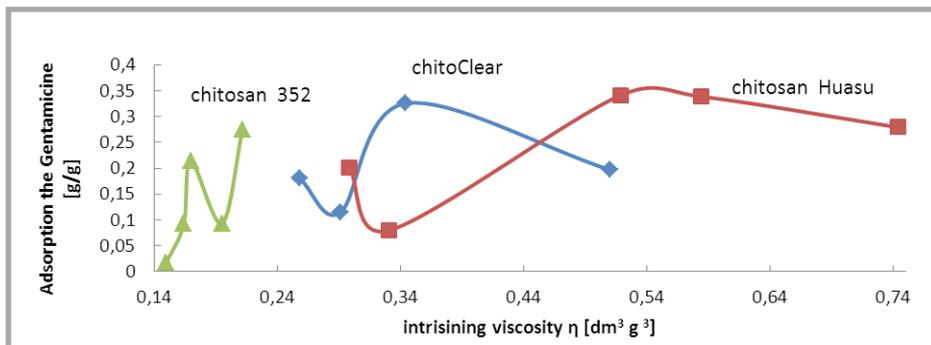
Examination of adsorption of the investigated drug: Gentamycin confirm that the amount of bounded drug depends on the degradation rate of chitosan and its origin. The analysis of the effect of intrinsic viscosity on the capability of absorption of the investigated drug by chitosan's reveals an increase in the amount of absorbed drug [2].

Regardless of the used chitosan an increase in the adsorption of the antibiotic on the chitosan with medium viscosity is noted. Low and high viscosity Chitosans are characterized by reduced adsorption, only chitosan type 352 for the variety of high viscosity has good adsorption properties. Presumably, this is due to the adsorption of particles with atoms with free electron pairs, and we have to expect a qualitative analogy of their specific adsorption to the amino groups of chitosan and probably there is the impact of a large dipole moment of the-NH<sub>2</sub> groups and blocking the active sites of the adsorbent surface by the solvent is observed.

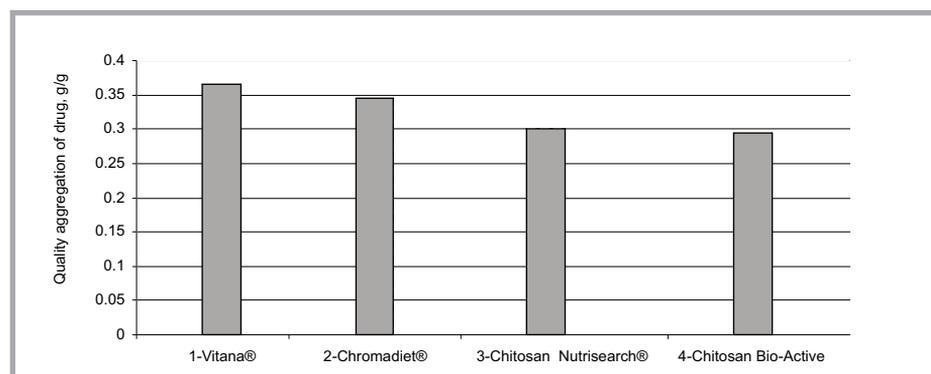
The analysis of viscosity-average molecular weights demonstrated that absorption of individual drug increases with decrease in intrinsic viscosity of all the investigated chitosans. The decrease in molecular weight induces lower strength of the polymer chain and the capability of formation of larger branches in the polymer network (*Figure 1* see page 178).

The binding of Gentamycin by chitosan's contained in dietary supplements confirms the hypothesis of aggregative character of chitosan in relation to these drug. The highest amounts of drug are bound by chitosan contained in Vitana® dietary supplement (*Figure 2* see page 178).

The investigated chitosan's contained in dietary supplements available on the weight-losing products market are capable of binding on an average of 0.292 – 0.371 g of Gentamycin per 1 g of chitosan end equivalent drug (0.06 g) . The highest absorption rate was observed at pH above 7.6 (intestinal juice).



**Figure 1.** The graph illustrating the dependence between binding the Gentamicin near dose 0.01 g in relation to essential intrinsic viscosity  $[\eta]$  (temp. 37 °C and pH 7.6).



**Figure 2.** The amount of Gentamicin bound by 1 g of chitosans present in dietary supplements in g/g (temp. 37 °C and pH 7.6).

## 4. Conclusion

In conclusion, the studies have shown that there is an interaction between drug and chitosan, what decreases their quantity and affects the bioavailability of these drugs. At pH above 7.6, corresponding to the environment of the intestinal contents, the mean absorption for the highest dose of the drug on chitosan ranged from 2.1 to 34.1%, what is associated with an increased essential surface of the polymer and its sorption capability. Basing on the above considerations, it can be stated that the investigated drug and polymer interact antagonistically in the form of absorption of drug from this group on a polymer such as chitosan.

## 5. References

1. Meler J., Pluta J., Ulanski P., Krotkiewski M.; Fat - the binding capacity of ninths - the modified and modified chitosans. In: *Progress he Chemistry and Application of Chitin and its Derivatives*. Vol. IX (ed.: H. Struszczyk). Polish Chitin Society. Lodz. pp. 129-136. 2003.

2. Roberts G.A.F. and Domszy J.G.; *Determination of the viscosimetric constants for chitosan*. *Int. J. Biol. Macromol.* 4; 374: 1982.
3. Mccurdy J.D.; *FDA and the use of chitin and chitosan derivative.*- In: *Advances in Chitin and Chitosan.. Elsevier Applied Science. London.* pp. 659-662.1992
4. Meler J., Pluta J., Krotkiewski M.; *The influence of various kinds of chitosan on fat binding ability.*4<sup>th</sup> *World Meeting on Pharmaceutics. Biopharmaceutics and Pharmaceutical Technology. Florence.* pp. 617-618 (2002).
5. Meler J., Pluta J., Ulański P., Krotkiewski M.; *Vozdejstvie raznyh form chitozana na sposobnost' svjazyvanija žirov. Modern perspectives in chitin and chitosan studies: Proceedings of the VII<sup>th</sup> International Conference. St. Petersburg - Repino.* pp. 258-260 *Moscow VNIRO Publishing* (2003).
6. Meler J., Pluta J.; *Influence on action of enzyme of feed wire in research chitosans in vitro.* In : *Biomaterials in Regenerative Medicine Vol. 6* (ed.: A. Nadolny) . *Polish Academy of Sciences. Vienna* pp. 173-177 (2006).
7. Meler J.; *Influence of different changes on bioavailability of medicine chitosans antiphlogistic drugs.* In: *Progress of Chemistry and Application of Chitin and its Derivatives. Vol. XIII* (ed.: A. Jaworska). *Polish Chitin Society. Łódź.* pp. 81-88 (2008).
8. Meler J., Pluta J.; *The effect of auxiliary substances the activity of lipase pancreatic biopharmaceutical patternelof digestive tract.* In: *Progress of Chemistry and Application of Chitin and its Derivatives. Vol. X* (ed.: H. Struszczyk), *Polish Chitin Society, Łódź,* pp. 131-137 (2004).
9. Grimling B., Meler J., Pluta J.: *Study of interaction of gastrointestinal agents in the presence of cytoprotective drug including bismuth W: Pierwiastki, środowisko i życie człowieka; Ed. Pasternak K.; Lublin: Polskie Towarzystwo Magnezologiczne, 2009; pp. 65-74.*
10. Meler J., Grimling B., Pluta J.: *Investigation on adsorption of fatty and bile acids in the presence of dietary supplements containing chromium J.Elementol. 2010 Vol.15 no. 1; pp. 141-147.*
11. Meler J.: *Influence of different change on bioavailability of medicine chitosans antiphlogistic drugs Progress on Chemistry and Application of Chitin and Its Derivatives 2008 Vol. 13; pp. 81-88.*
12. Meler J.: *The effect of physicochemical factors on absorption properties of certain spasmolytics in the presence of dietary supplements containing chitosan. Progress on Chemistry and Application of Chitin and Its Derivatives 2009 Vol. 14; pp.133-143.*

